Fe3+和 Fe2+对原代培养的成骨细胞增殖、分化和矿化功能的影响

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摘要:采用噻唑蓝(MTT)法、碱性磷酸酶(ALP)比活性测定、油红 O 染色和茜素红染色及定量分析,研究了不同浓度的 Fe^{3+} 和 Fe^{2+} 对原代培养的成骨细胞增殖、分化及矿化功能的影响。结果表明:浓度为 $1\times10^{-9}\sim1\times10^{-4}$ mol·L⁻¹ 的 Fe^{3+} 和 Fe^{2+} 促进成骨细胞增殖,但是在较高浓度 1×10^{-3} mol·L⁻¹ 时,它们则抑制成骨细胞增殖。与成骨细胞作用 48 h,浓度为 $1\times10^{-8}\sim1\times10^{-4}$ mol·L⁻¹ 的 Fe^{3+} 和 Fe^{2+} 抑制其分化,但在较低的浓度 1×10^{-9} mol·L⁻¹ 时则对其分化没有影响;进一步延长作用时间为 72 h, Fe^{3+} 对成骨细胞分化没有影响,除 1×10^{-6} mol·L⁻¹ 浓度的 Fe^{2+} 促进成骨细胞分化外,其他浓度的 Fe^{2+} 则抑制其分化;测试浓度下的 Fe^{3+} 对成骨细胞向脂肪细胞的横向分化表现为抑制或没有影响,而 Fe^{2+} 的影响则依赖于浓度和作用时间。在 $1\times10^{-8}\sim1\times10^{-5}$ mol·L⁻¹ 浓度范围内, Fe^{3+} 和 Fe^{2+} 对矿化结节的影响表现出相反的效应。在较高浓度(1×10^{-4} mol·L⁻¹)下,它们促进矿化节结的形成,而在较低浓度(1×10^{-9} mol·L⁻¹)下, Fe^{3+} 抑制矿化节结的形成, Fe^{2+} 则没有影响。结果提示:浓度,作用时间和铁离子的价态都是影响 Fe^{3+} 和 Fe^{2+} 生物效应(从毒性到活性,从损伤到保护,从上调到下调)转变的关键因素。

关键词:铁离子;成骨细胞;增殖;分化;矿化

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Effects of Fe³⁺ and Fe²⁺ on Proliferation, Differentiation and Mineralization Function of Primary Osteoblasts *in vitro*

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Abstract: The 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromid (MTT), alkaline phosphatase (ALP) activity, oil red O assays and alizarin red-S (ARS) stain were used to evaluate the effects of Fe³⁺ and Fe²⁺ on proliferation, differentiation and mineralization function of primary osteoblasts (OBs) in vitro. The results indicate that both Fe³⁺ and Fe²⁺(1×10⁻⁹~1×10⁻⁴ mol·L⁻¹) promote the proliferation of OBs, but turn to inhibit at a higher concentration of 1×10^{-3} mol·L⁻¹. Both Fe³⁺ and Fe²⁺(1×10⁻⁸~1×10⁻⁴ mol·L⁻¹) inhibit differentiation of OBs, but have no effect on differentiation at a lower concentration of 1×10^{-9} mol·L⁻¹ for 48 h. Whereas, Fe³⁺ shows no effect on differentiation of OBs, Fe²⁺ begins to promote differentiation of OBs at concentration of 1×10^{-6} mol·L⁻¹, but Fe²⁺

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inhibits differentiation of OBs at other concentrations by prolonging the incubation time. Fe³⁺ inhibits adipocytic trans-differentiation of OBs at tested concentrations. The promotion or inhibition effect of Fe²⁺ on adipocytic trans-differentiation of OBs depends on concentration and incubation time. Fe³⁺ and Fe²⁺($1 \times 10^{-8} \sim 1 \times 10^{-5}$ mol·L⁻¹) inhibit and promote the formation of mineralized matrix nodules of OBs, respectively. Fe³⁺ and Fe²⁺ promote the formation of mineralized matrix nodules at a higher concentration of 1×10^{-4} mol·L⁻¹, but Fe³⁺ inhibits the formation of mineralized matrix nodules, Fe²⁺ shows no effect at a lower concentration of 1×10^{-9} mol·L⁻¹. The results suggest that concentration, culture time and valence state of iron ion are key factors for switching the biological effects of Fe³⁺ and Fe²⁺ from toxicity to activity, from damage to protection, or from down-regulation to up-regulation.

Key words: iron ion; osteoblasts; proliferation; differentiation; mineralization

0 Introduction

As a common metabolic bone disease, osteoporosis (OP) is characterized by low bone mass, a disturbance of bone micro-architecture, the increasing incidence of bone fragility and fracture^[1]. With the coming of aging society, osteoporosis occurs more and more frequently, severely impairs the quality of life, increases mortality and care costs ^[2–5]. It has been recognized as a major public health problem and much attention has been focused on searching powerful medicine and therapeutic strategy.

In the whole life, bone remodeling occurs continuously to maintain the normal bone mass, in which osteoclasts (OCs) and osteoblasts (OBs) are two major cells involved^[6,7]. The abnormality in the balance between OB and OC activities results in osteoporosis. A great deal of evidence suggests that osteoporosis is the result of a reciprocal interaction between genetic susceptibility and environmental factors^[8]. Among them, certain essential trace elements were reported to be involved in the pathogenesis of osteoporosis [9~14]. As an essential trace element, adequate iron plays irreplaceable roles in body activities. Iron loaded persons are at increased risk for infection, neoplasia, arthropathy, cardiomyopathy, an array of endocrine, neurodegenerative diseases and osteoporosis^[15~17]. Animal experiment has shown that the iron content is significantly lower than control group in rats with resection of ovarian osteoporosis model^[18]. Low molecular mass iron chelators as well as a natural protein iron chelator, lactoferrin, may be useful in prevention of osteoporosis ^[19]. It has been suggested that iron-deficient rats have lower bone mass than iron-replete animals ^[20]. However, the effects of Fe³⁺ and Fe²⁺ on bone-forming function of primary OBs in vitro were not reported so far. In order to elucidate the effects of Fe³⁺ and Fe²⁺ on bone metabolism at cell levels, the effects of Fe³⁺ and Fe²⁺ on the proliferation, differentiation and mineralization function of primary OBs in vitro were studied in this paper.

1 Materials and methods

1.1 Materials and reagents

Institute of Cancer Research (ICR) mice were obtained from Agricultural University of Hebei. Dulbeccos modified Eagles medium (DMEM) and trypsin were purchased from GIBCO, USA. 3-(4,5dimethyl-2-thiazolyl)-2,5-diphenyl-2H tetrazolium bromide (MTT), penicillin and streptomycin, β -glycerphosphate, dexamethasone, ascorbic acid, collagenase type II, insulin, oil red O stain, alizarin red-S(ARS) and cetylpyridium chloride were from Sigma, USA. Neonatal bovine serum (NBS) was from Hangzhou Sijiqing Organism Engineering Institute. Alkaline phosphatase (ALP) activity kit was obtained from Nanjing Jiancheng Biological Engineering Institute (Nanjing, China), and micro-protein assay kit was purchased from Beyotime Biotechnology (Haimen, China). Ferric chloride and Ferrous chloride (Purity>99.9%) were analytical reagent

(AR).

1.2 Methods

1.2.1 Isolation and culture of primary OBs

OBs were isolated mechanically from newborn ICR mouse skull by sequential digestion as reported previously with minor modification $^{[21]}$. Briefly, skull was dissected, endosteum and periosteum were stripped off, and the bone was cut into approximately $1 \sim 2~\text{mm}^2$ pieces and digested with trypsin(2.5 g · L $^{-1}$) for 30 min and the digestion was discarded. Then the bone was digested with collagenase II (1.0 g · L $^{-1}$) twice with 1 h for each, and the cells were collected and cultured. After incubating overnight at 37 °C , in a 5% CO2 humidified incubator, the DMEM was removed. Then the medium was replaced every 3 d in all experiments.

1.2.2 Cell proliferation assay

The protocol described by Mosmann was followed with some modifications^[22]. Briefly, OBs(2×10⁴ cells per well) were plated in 96-well culture plates and cultured overnight at 37 °C, in a 5% CO₂ humidified incubator. Fe3+ or Fe2+ was added to the wells at various concentrations. Control wells were prepared by addition of DMEM. Wells containing DMEM without cells were used as blanks. Wells containing NaF (1×10⁻⁶ mol·L⁻¹) without Fe³⁺ or Fe²⁺ were used as positive control. Upon completion of the incubation, MTT dye solution (20 µL, 5 mg·mL⁻¹) was added to each well. After 4 h incubation, the supernatant was removed and DMSO (100 µL) was added to solubilize the MTT. The optical density(OD) of each well was measured on a microplate spectrophotometer (BioRad Model 3550, USA) at a wavelength of 570 nm. The proliferation rate (%) was calculated according to the formula: [(OD_{treated}-OD_{control})/ $OD_{control} \times 100\%$.

1.2.3 ALP activity assay

OBs (2×10⁴ cells per well) were plated in 48-well culture plates, and treated with Fe³⁺ or Fe²⁺ for 48 h and 72 h, respectively. The plates were washed thrice with ice-cold PBS and lysed by two cycles of freezing and thawing. Aliquots of supernatants were subjected to ALP activity and protein measurement using an ALP kit and a micro-protein assay kit, respectively. All results were normalized by protein content.

1.2.4 Oil red O stain and measurement

The $OBs(3\times10^4 \text{ cells per well})$ were plated in 48well culture plates, after being induced by adipogenic supplement(10 µg·mL⁻¹ insulin, 10⁻⁷ mol·L⁻¹ dexthamethone) and treated with Fe³⁺ or Fe²⁺, and fat droplets within differentiated adipocytes from OBs were stained using the oil red O described by Ichiro et al. [23] with some modifications. Briefly, cell monolayers were fixed in 4% formaldehyde, washed in water and stained with a 0.6%(W/V) oil red O solution(60% isopropanol, 40% water) for 15 min at room temperature. For quantification, cell monolayers were then washed extensively with water to remove unbound dye, and recorded by inverted phase contrast microscopy (Olympus IX 51), then 1 mL of isopropyl alcohol was added to the culture plates. After 5 min, the absorbance of the extract was measured by a spectrophotometer at 510 nm. The adipocytic trans-differentiation promoting rate (%) was calculated according to the formula: [(OD_{treated}-OD_{control})/ $OD_{control} \times 100\%$.

1.2.5 Mineralized matrix formation assay

OBs(3×10⁴ cells per well) were plated in 24-well tissue culture plates and cultured overnight at 37 °C, in a 5% CO₂ humidified incubator. The medium was then changed to differentiation medium containing 10 mmol. $L^{-1}\beta$ -glycerophosphate and 50 $\mu g \cdot mL^{-1}$ ascorbic acid in the presence or absence of Fe^{3+} or Fe^{2+} for 21 d. The formation of mineralized matrix nodules was determined by ARS stain. Briefly, Cell monolayers were fixed in 95% ethanol for 10 min at room temperature, then washed by PBS twice and stained with 0.1% alizarin red S for 30 min also at room temperature. Quantitation of ARS staining was performed by elution with 10%(W/V)cetylpyridium chloride for 10 min at room temperature and the absorbance was measured at 570 nm^[24]. Results were expressed as moles of ARS/per milligram of total cellular protein and the mineralized matrix promoting rate(%) was further calculated.

1.2.6 Statistical analysis

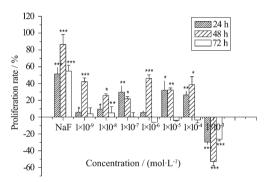
Data were collected from at least four separate experiments. The results were expressed as means \pm standard deviation(SD). The statistical differences were analyzed using SPSS' t-test. P values less than 0.05

were considered to be statistically different.

2 Results

2.1 Effects of Fe³⁺ and Fe²⁺ on the proliferation of OBs

As shown in Fig.1 and Fig.2, Fe³+(1×10 $^{-9}$ ~1×10 $^{-4}$ mol·L $^{-1}$) promotes the proliferation of OBs for 24 h and 48 h, but the influence of Fe³+ on the proliferation disappears for 72 h. Fe²+(1×10 $^{-9}$ ~1×10 $^{-4}$ mol·L $^{-1}$) promotes the proliferation of OBs for 24 h, 48 h and 72 h. Fe³+ and Fe²+ inhibit the proliferation of OBs at a higher concentration of 1×10 $^{-3}$ mol·L $^{-1}$.



*P < 0.05, **P < 0.01, ***P < 0.001 vs control group, n = 5

Fig.1 Effect of Fe³⁺ on the proliferation of OBs

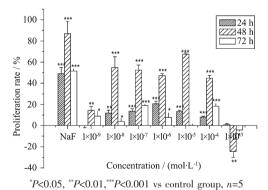
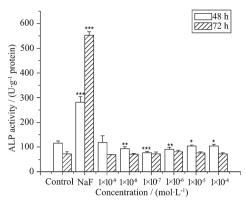


Fig.2 Effect of Fe²⁺ on the proliferation of OBs

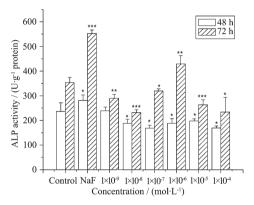
2.2 Effects of Fe³⁺ and Fe²⁺ on the differentiation of OBs

As shown in Fig.3 and 4, after incubating 48 h, both Fe³⁺ and Fe²⁺($1 \times 10^{-8} \sim 1 \times 10^{-4} \text{ mol} \cdot L^{-1}$) inhibit differentiation of OBs, but show no effect on differentiation at a lower concentration of 1×10^{-9} mol $\cdot L^{-1}$. Whereas, Fe³⁺ has no effect on differentiation of OBs, Fe²⁺ turns to promote differentiation of OBs at concentration of 1×10^{-6} mol $\cdot L^{-1}$, but inhibits differentiation of



*P<0.05, **P<0.01, ***P<0.001 vs control group, n=5

Fig.3 Effect of Fe³⁺ on the differentiation of OBs



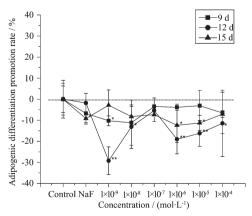
*P<0.05, **P<0.01, ***P<0.001 vs control group, n=5

Fig.4 Effect of Fe²⁺ on the differentiation of OBs

OBs at other concentrations by prolonging the incubation time.

2.3 Effects of Fe³⁺ and Fe²⁺ on the adipocytic transdifferentiation of OBs

As shown in Fig.5, Fe³⁺ inhibits adipocytic transdifferentiation of OBs at concentrations of 1×10^{-8} and

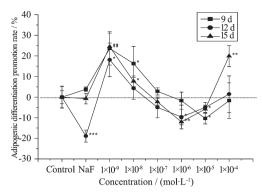


*P<0.05, **P<0.01, ***P<0.001 vs control group, n=5

Fig.5 Effect of Fe $^{3+}$ on the adipocytic trans-differentiation of OBs

 $1\times10^{-9}\,\mathrm{mol}\cdot\mathrm{L}^{-1},$ while it shows no effect at other tested concentrations for 9 d. Fe³+ inhibits adipocytic transdifferentiation of OBs at all tested concentrations except $1\times10^{-7}\,\mathrm{mol}\cdot\mathrm{L}^{-1}$ for 12 d. Fe³+ inhibits adipocytic transdifferentiation of OBs at concentrations of $1\times10^{-6},\ 1\times10^{-5}\,\mathrm{mol}\cdot\mathrm{L}^{-1},$ while has no obvious effect at other tested concentrations for 15 d.

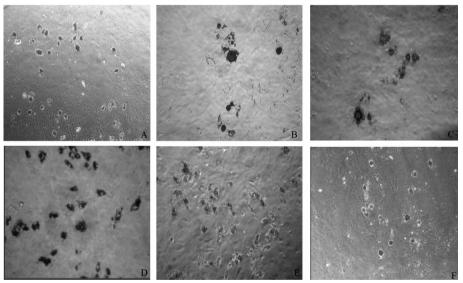
As shown in Fig.6, Fe²⁺ promotes adipocytic trans-differentiation of OBs at concentrations of 1×10^{-9} and 1×10^{-8} mol·L⁻¹, turns to inhibit at concentration of 1×10^{-5} mol·L⁻¹, then has no effect at a higher concentration of 1×10^{-4} mol·L⁻¹ for 9 d. Fe²⁺ promotes adipocytic trans-differentiation of OBs at concentration of 1×10^{-9} mol·L⁻¹, turns to inhibit at concentrations of 1×10^{-6} and 1×10^{-5} mol·L⁻¹, then shows no effect at a higher concentration of 1×10^{-4} mol·L⁻¹ for 12 d. Fe²⁺ promotes adipocytic trans-differentiation of OBs at concentrations of 1×10^{-9} and 1×10^{-8} mol·L⁻¹, inhibits



*P < 0.05, **P < 0.01, ***P < 0.001 vs control group, n = 5

Fig.6 Effect of Fe²⁺ on the adipocytic trans-differentiation of OBs

adipocytic trans-differentiation of OBs at concentrations of 1×10^{-6} and 1×10^{-5} mol·L⁻¹, then turns to promote adipocytic trans-differentiation at a higher concentration of 1×10^{-4} mol·L⁻¹ for 15 d. The experimental results are in accordance with morphological observations(Fig.7).



(A) Control group; (B) $1 \times 10^{-6} \text{ mol} \cdot \text{L}^{-1} \text{ NaF}$; (C) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{3+}$; (D) $1 \times 10^{-7} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{3+}$; (E) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-7} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ Fe}^{2+}$; (F) $1 \times 10^{-8} \text{ mol} \cdot \text{L}^{-1} \text{ mol} \cdot \text{L}^{-1} \text{ mol} \cdot \text{L}^{-1} \text{ mol} \cdot \text{L}^$

Fig.7 Adipocytes stained by oil red O when OBs being cultured in the presence of adipogenic supplement for 12 d

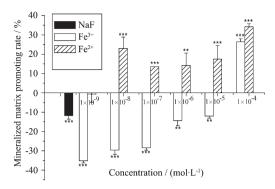
2.4 Effects of Fe³⁺ and Fe²⁺ on the formation of mineralized matrix nodules

As shown in Fig.8, Fe³⁺($1 \times 10^{-9} \sim 1 \times 10^{-5}$ mol·L⁻¹) inhibits the formation of mineralized matrix nodules of OBs, but promotes the formation of mineralized matrix nodules at a higher concentration 1×10^{-4} mol·L⁻¹. Fe²⁺ ($1 \times 10^{-8} \sim 1 \times 10^{-4}$ mol·L⁻¹) promotes the formation of mineralized matrix nodules, but shows no effect at a

lower concentration 1×10⁻⁹ mol·L⁻¹.

3 Discussion

In present study, an *in vitro* model was used to examine the effects of Fe³⁺ and Fe²⁺ on the proliferation, differentiation and mineralization function of primary OBs. The results indicate that the effects of Fe³⁺ and Fe²⁺ on proliferation, differentiation and mineralization



*P < 0.05, **P < 0.01, ***P < 0.001 vs control group, n = 5

Fig.8 Effects of Fe^{3+} and Fe^{2+} on the mineralized nodule formation of OBs

function of primary OBs depend on concentration and incubation time, but are not dose-dependent. Many trace elements have been reported to be involved in the pathogenesis of OP. Cen et al. [25,26] have reported that Zn2+ could promote proliferation and differentiation of the rat OBs in vitro. Zn2+ could promote the transition of cell cycle from phases G0/G1 to S phase, thereby promote DNA synthesis and cell proliferation. It could activate the phosphoinositol signal transduction system in rat OBs in vitro, and elevate triphosphate inositol (IP3) content and protein kinase C(PKC) activity. The mechanism includes not only the membrane receptormediated pathway but also possibly the direct Zn² + action. Fu et al. [27] reported Gd3+(2~60 µmol·L-1) exerted proliferation-promoting effects in a concentrationdependent manner. Gd^{3+} had remarkable effect on cell cycle distribution. Gd3+(60 µmol·L-1) drove the synchronized cells to enter S phase at 12 h and G2/M phase at 18 h by increasing the levels of cyclins D, E, A and phosphorylated retinoblastoma protein and decreasing the expressions of p21^{Cipl} and p27^{Kipl}. Gd³⁺ drove cells through G1/S transition point and promoted cell cycle progression via activation of both ERK and PI3K signaling pathways. It was reported that H₂O₂ promoted osteoblastic differentiation of vascular smooth muscle cells by inducing oxidative stress in vascular smooth muscle cells and subsequently activating MAPK pathways. Shi et al.^[28] reported that La³⁺ suppressed the effect of H₂O₂ probably by meliorate cell redox status through MAPK pathway. So we deduce that Fe³⁺ and Fe²⁺ promote proliferation of OBs probably by promoting the transition of cell cycle from phases G0/G1 to S phase thereby promoting DNA synthesis and cell proliferation. The difference between Fe 3 + and Fe 2 + might be related to reactive oxygen species.

Medullary adipocytes are secretory cells that may influence hematopoiesis and osteogenesis. Adipocytic and osteogenic cells are believed to be derived from multipotential stromal cells in the marrow, and in vitro studies have shown an inverse relationship between the differentiation of adipocytic and osteogenic cells [29]. It was reported that preadipocytes isolated from mouse marrow might regulate the activity and final differentiation of OBs. The condition medium harvested from mouse stromal preadipocytes decreased the ALP activity of a mouse stromal osteoblastic cell line [30]. Some investigators suggested that adipocytes might be involved in hematopoietic and osteogenic process by supplying the necessary soluble cell surface factors for OC differentiation and function in vitro [31]. Sakaguchi et al. [32] demonstrated that adipocyte-enriched stromal cells supported OC formation. Benayahu et al. [33] reported that preadipocytes also had the potential to stimulate OC differentiation. Adipocytes synthesized and released a variety of peptide and nonpeptide compounds or secreted cytokines such as tumor necrosis factor- α (TNF- α) and interleukin(IL-6), and the main effect of these cytokines is a stimulation of bone resorption^[34]. So a reversal of adipogenesis will provide an important therapeutic approach to prevent agedrelated and steroids-induced osteoporosis. experimental results indicate that the effect of Fe³⁺ and Fe^{2 +} on adipocytic trans-differentiation of OBs also depends on concentration and incubation time. Fe3+ and Fe2+ may have protective effect on bone and this protective effect on bone may be mediated by inhibiting adipocytic trans-differentiation of OBs which may promote differentiation and mineralization of OBs.

In conclusion, the effects of Fe^{3+} and Fe^{2+} on the proliferation, differentiation and mineralization function of primary OBs *in vitro* are very complicated, but concentration, culture time and valence states of iron ion are key factors for switching the biological effects of Fe^{3+} and Fe^{2+} from toxicity to activity, from damage to

protection, or from down-regulation to up-regulation. The mechanism of the effects of Fe³⁺ and Fe²⁺ on the proliferation, differentiation, adipocytic transdifferentiation and mineralization function of primary OBs remains to be further studied.

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