

氧化铟锡(ITO)自组装修饰及其对有机电致发光器件性能的影响

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摘要: 设计合成了一种新型的有机硅氧烷 Cz-Si, 并将其用于 ITO 自组装修饰。制备的 Cz-Si 具有较好的稳定性, 可以在空气对 ITO 进行自组装修饰, 实验操作简单。为考察 ITO 自组装修饰对有机电致发光器件性能的影响, 分别以修饰后的 ITO(ITO/SAM)及不修饰的 ITO(unmodified)作阳极, 制备了一系列有机电致发光器件 ITO/SAM(or unmodified)/NPB(40~50 nm)/Alq₃(60 nm)/LiF(1.0 nm)/Al。实验结果表明, ITO 自组装修饰后器件性能可以得到显著提升, 研究认为这与其调控 ITO/有机层界面的电子能级、粗糙度以及界面一致性有关。

关键词: ITO 修饰; 自组装; 有机电致发光器件

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Indium Tin Oxide Anode Self-Assembled Monolayer Modification for Device Performance Improvement of Organic Light-Emitting Diodes

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Abstract: A novel alkoxy silane (Cz-Si) was synthesized and used as an active reagent for self-assembled monolayer (SAM) modification on indium tin oxide (ITO) surface. The as-prepared Cz-Si could modify the ITO surface successfully under the mild atmosphere without any protection, offering simple and easy experimental operation. To investigate the effect of SAM modification on the device performances, based on ITO/SAM anodes, a series of organic light-emitting diodes (OLEDs) were fabricated: ITO/SAM (or unmodified)/NPB (40~50 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al. The devices show improvement compared to their counterparts with bare ITO anodes. The improvement could be attributed to the modulating of the electronic energy, surface roughness and interface integrity at the ITO/hole transporting layer (HTL) interface by SAM modification.

Key words: ITO modification; self-assembled monolayer (SAM); organic light-emitting diodes (OLEDs)

0 Introduction

Interfaces play a pivotal role for the performance of the organic light-emitting diodes (OLEDs). Anode/hole transportation layer (HTL) is one of the most

important interfaces, which determines the hole injection efficiency. Indium tin oxide (ITO) has been the most commonly used anode in OLEDs due to its high conductivity and transparency. Tailoring of ITO surface properties represents a highly desirable

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method to improve the device performance. There are many physical and chemical treatments to improve the ITO/HTL interface characteristics, aiming at modulating the work function of the electrode, tuning the balance of carrier injection, smoothening the surface roughness, and eliminating the disparity of ITO/HTL interface, such as O_2 plasma^[1], UV ozone^[2], chlorination^[3], or coating the ITO with a buffer layer, such as copper phthalocyanine (CuPc)^[4-5], poly 3,4-ethylene dioxythiophene: polystyrene sulfonate: (PEDOT:PSS)^[6-7], CHF_x ^[8-9], etc.. Insertion of transition metal oxide thin film between the ITO and the organic layer has also been another efficient way to improve the carrier injection, such as MnO ^[10], MoO_3 ^[11], Al_2O_3 ^[12], and WO_3 ^[13].

Although there are numerous ways to improve the surface properties of ITO, one of the most successful modifications is the use of monolayers based on organic molecules with widely variable and functional groups. Introducing a self-assembled monolayer (SAM) on ITO surface allows the molecular engineering of the ITO/organic interface, and results in the fine control of the interfacial chemistries and electronics. In particular, both surface energy matching and work function compatibility at ITO/HTL interface can be remarkably improved. For the formation of robust monolayers on ITO, a variety of different functional groups can be used, such as phosphonic acids^[14], carboxylic acids^[15], thiols, alcohols, and organosilanes (e.g., alkoxy- and chlorosilanes)^[16-17]. Organosilanes have found great utility in the modification of metal oxides in general^[18-19]. Many groups have shown their brilliant work on this field. Huang Q. et al^[20-21] achieved remarkable enhancement of the device performances by ITO SAM modification, but the organosilane (e.g., chlorosilanes) they used were highly sensitive to moisture^[16-21], and the modification experiments should be operated under rigorous protection, which makes experimental operation complex.

In this work, a novel alkoxysilane Cz-Si was synthesized and used as an active reagent for SAM modification on ITO surface. The synthesized Cz-Si is stable in air, and the SAM modification operation could be processed without any protection. A series of

devices were fabricated with ITO/SAM as anodes, and the effect of SAM on the device performance was investigated. The device performance could be improved significantly by ITO Cz-Si-SAM modification.

1 Experimental

1.1 Materials and instruments

9-ethyl-9*H*-carbazol-3-amine (95%) and triethoxy (3-isocyanatopropyl)silane (95%) were purchased from Sigma-Aldrich. NPB and Alq_3 (99.99%) were purchased from Beijing Yilaite Optoelectronic Materials Company.

1H NMR spectra were recorded on a Mercury Plus-400 spectrometer in $DCCl_3$. Cyclic Voltammetry curve of Cz-Si was measured by CHI600B analyzer (Shanghai Chenhua Instruments Company). XPS data were taken by Thermo ESCALAB 250.

1.2 Synthesis of Cz-Si

1.05 g 9-ethyl-9*H*-carbazol-3-amine (5.0 mmol) was added into a 100 mL three-necked flask, and dissolved by 30 mL anhydrous chloroform. The solution of 1.33 mL (5.0 mmol) triethoxy(3-isocyanatopropyl)silane in 10 mL anhydrous chloroform was dropped into the reaction flask slowly under stirring. The reaction mixture was refluxed for 12 h, and then most of the excess solvent was evaporated, and the remainder was cooled to room temperature. After that the rest reaction solution was poured into 300 mL petroleum ether, and a large amount of white precipitate appeared. After filtration, the precipitate was washed with petroleum ether several times, and dried in vacuum. The product (named as Cz-Si) was 1.93 g (yield 84%). The purification of the crude product: the product was dissolved in a little of chloroform again, and a large amount of petroleum ether was added, and then the precipitate was washed and dried. 1H NMR($CDCl_3$, 400 MHz): δ 0.61 (t, 2H), 1.15 (t, 9H), 1.22 (q, 5H), 3.24 (broad, 2H), 3.73 (m, 8H), 7.2~7.5 (m, 7H), 7.98 (broad, 1H), 8.07 (1H).

1.3 Characterization of the electrochemical property of Cz-Si

Cz-Si was dissolved in $CHCl_3$ ($1.0\text{ mmol}\cdot\text{L}^{-1}$). $n\text{-Bu}_4\text{N}^+(\text{ClO}_4)^-$ solution ($0.1\text{ mol}\cdot\text{L}^{-1}$ in $CHCl_3$) was used as the supporting electrolyte. Cyclic voltammetry (CV)

characterization was carried out on a CHI 600B electrochemical workstation with a Pt flake working electrode, a Pt wire counter electrode and a saturated calomel reference electrode.

1.4 ITO cleaning

The ITO substrates were first cleaned by detergent solution (FRANKLAB S. A company), and then ultrasonicated for 10 minutes in ethanol, acetone, isopropanol, separately. Finally the substrates were dried and then used for device fabrication or SAM modification directly.

1.5 SAM Modification of the ITO surface

The cleaned ITO substrates were soaked in 10 mmol·L⁻¹ Cz-Si solution in chloroform at room temperature (RT) for 0~96 h. The ITO substrates were washed with chloroform, and then ultrasonicated for 10 min. in chloroform and acetone: DI water (100:1, V/V) solution successively to remove any physisorbed residues. Finally, the Cz-Si-modified ITO substrates were dried at 120 °C for 30 min.

1.6 Characterization of ITO-SAM

The bare ITO (cleaned by routine procedure) and the Cz-Si modified ITO (ITO-SAM) were characterized by XPS and aqueous contact angle. The wettability of the ITO and SAM-ITO surface was characterized by static contact angle measurement at room temperature with DI water as medium. The volume of the water droplet was 2 μL, and the data were recorded when the water droplet maintained for 5 s, and the final result was the average value of ten measurements.

1.7 Fabrication of OLEDs

N, N'-di(naphthalen-1-yl)-*N, N'*-diphenyl-[1, 1'-biphenyl]-4,4'-diamine (NPB) was used as the hole transport layer. Tris(8-quinolinolato) aluminum(III) complex (Alq₃) was used as the emitting layer and electron transport layer. LiF/Al was used as the cathode. Organic films were deposited onto ITO substrates successively by thermal evaporation in a vacuum chamber $\leq 2 \times 10^{-4}$ Pa, followed by a 1.0 nm LiF, and then 150-nm-thick Al cathode. The film thicknesses and the deposition rate were monitored by quartz oscillators. Deposition rates were ~ 0.1 nm·s⁻¹ for organic films, ~ 0.01 nm·s⁻¹ for LiF films, and ~ 1

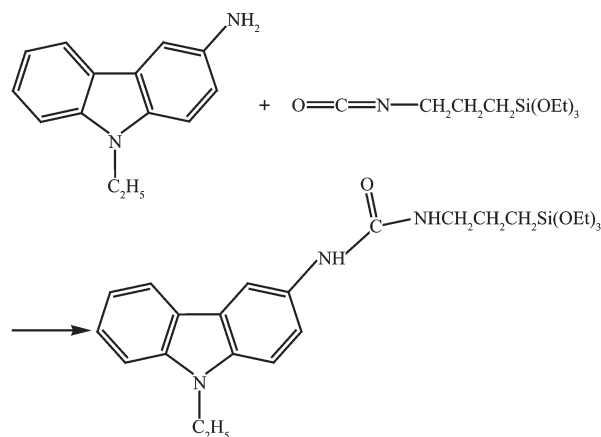
nm·s⁻¹ for the Al cathode, separately. The emitting area was 3×3 mm².

1.8 OLEDs measurements

Voltage-Current density-Brightness (*V-J-B*) characteristics were measured using a Keithley source measurement unit (Keithley 2400 and Keithley 2000) with a calibrated silicon photodiode. All measurements were carried out in ambient atmosphere at room temperature.

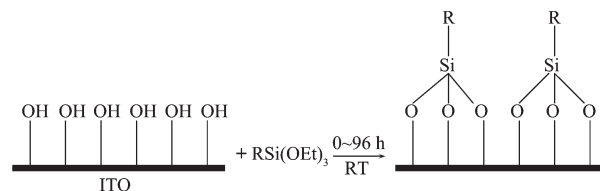
2 Results and discussion

The active material Cz-Si was synthesized according to the procedure shown in Scheme 1.



Scheme 1 Synthesis route for compound Cz-Si

The prepared Cz-Si contains a carbazole fragment, which will benefit for the hole transporting, and a reactive terminal group -Si(OEt₃), rendering the possibility for chemical modification on the ITO surface. When ITO is soaked in the solution of Cz-Si/CHCl₃, the -Si(OEt₃) group of Cz-Si would react with -OH group on the ITO surface, thus resulting in a self-assembly monolayer on the ITO surface. The SAM modification procedure is given in Scheme 2:



Scheme 2 Procedure for SAM modification of ITO (RSi(OEt)₃=Cz-Si)

Compared with -SiCl₃ group, which could react with the OH group on ITO rapidly^[16-19], the reaction

activity of the group $-\text{Si}(\text{OEt}_3)$ is relatively lower, thus the SAM modification procedure with Cz-Si should take relatively long time at RT, or need higher temperature. Although there is a little time-consuming, it renders the advantage that the SAM modification could be carried out at ambient atmosphere without any protection, offering the facilitation for the modification operation. In this work we investigated the effect of different periods of processing time (0~96 h) on the SAM at RT. It would lead to an undesirable SAM on ITO while the processing time is too short or too long, i.e., the coverage of SAM on ITO would be imperfect while it is too short and the self-condensation (due to the favorable energy of formation for the Si-O-Si linkage) would take place while it is too long. The modified ITO with a certain period SAM processing time is denoted as ITO/SAM (x) h in the following text.

To confirm that Cz-Si is chemically grafted onto the ITO surface, the bare ITO and ITO/SAM (24) h surfaces were characterized by XPS (Fig.1). As shown in Fig.1, the signal of $\text{In}3d_{3/2, 5/2}$ and $\text{Sn}3d_{3/2, 5/2}$ decreases after SAM modification, while that of $\text{Si}2s, 2p$ increases obviously, which is resulted from the coverage of Cz-Si on the ITO surface, indicating successful grafting of Cz-Si onto ITO surface, because all the physically absorbed residues are removed.

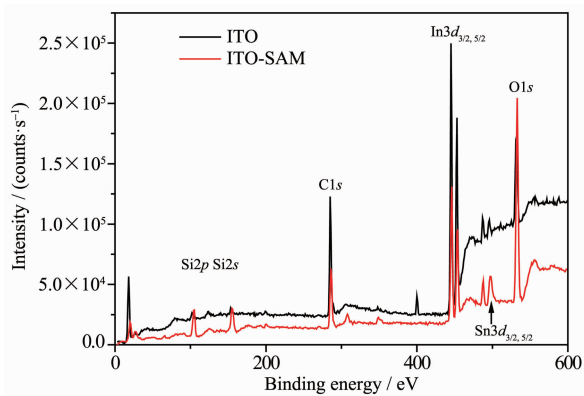


Fig.1 XPS spectra of the bare ITO (black line) and SAM modified ITO (red line)

ITO is highly hydrophilic with a heavily hydroxylated inorganic surface, while the organic HTLs (e.g., TPD and NPB) are hydrophobic. This huge difference is disadvantageous for the growth and

stability of the organic film. It is argued that a primary failure mechanism arises from significant mismatches of interfacial surface energy within the heterostructures^[22]. This large disparity in surface energies promotes the loss of ITO/HTL interfacial integrity, leading to chemical degradation and erosion of emissive efficiency of the device. So tuning the surface energy of the ITO (qualitatively indexed by advancing aqueous contact angle data) would result in positive effect on the performance of the OLEDs. Insertion of SAM between ITO and HTL layer would increase the hydrophobicity of the ITO surface, thus eliminating the inorganic/organic interfacial disparity at the ITO/HTL interface. The contact angles for bare ITO and ITO/SAM (24) h in our work are 31° and 73° , respectively (Fig.2). This result indicates that the SAM increases the hydrophobicity of ITO surface remarkably, and offers an integrated ITO/organic layer interface. The contact angle is reasonably expected to be higher when the soaking time of ITO in the Cz-Si solution increases, because it increases the coverage ratio of the Cz-Si on ITO.

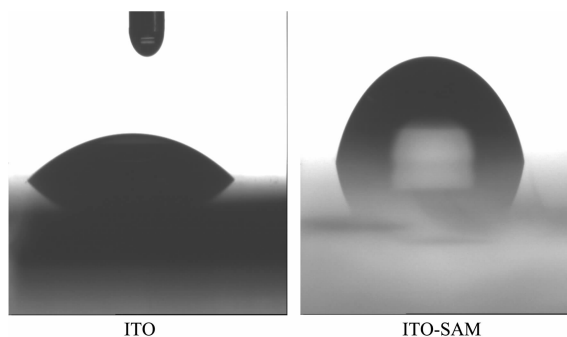


Fig.2 Aqueous contact angle of bare ITO (left) and SAM modified ITO (right)

The energy level of the active material Cz-Si is crucial to the hole injection of the device. The HOMO level of Cz-Si by CV is shown in Fig.3. The oxidation potential (E^{ox}) is 1.3 eV. According to the equation^[23]: $\text{HOMO} = -(E^{\text{ox}} + 4.8)$ eV, the calculated HOMO level of Cz-Si is -6.1 eV. In fact, the HOMO level of Cz-Si is lower than that of NPB (-5.4 eV)^[24-25], which is disadvantageous for the hole injection. Considering that the electron transporting rate of the electron transporting layer is lower than the hole transporting

rate of the hole transporting layer, the restriction of hole injection to some extent will benefit for the balance of the carriers, which is pivotal for the enhancement of the device performances and stability.

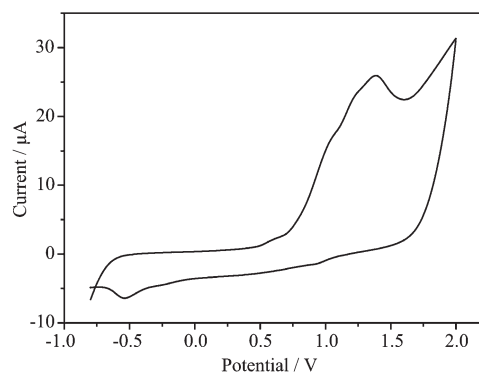


Fig.3 Cyclic voltammetry curve of Cz-Si

To investigate the effect of Cz-Si-SAM on the device performances, a series of devices with the classical structure were fabricated:

Device I : ITO/SAM (x) h/NPB (50 nm)/Alq₃ (60 nm)/LiF/Al

Device II : ITO/NPB (50 nm)/Alq₃ (60 nm)/LiF/Al

where x is the SAM processing time. When $x > 20$, Device I shows remarkable performance improvement compared with Device II. While when $x \leq 12$, the device exhibits similar performance to that of the unmodified counterpart. When $x = 38$, the device achieves the best performance. The turn-on voltage (V_{on}), maximum brightness (B_{max}) and maximum current efficiency (Eff_{max}) of this device are 5.5 V, 21 464 cd·m⁻² and 5.0 cd·A⁻¹, respectively, while that of the unmodified device (Device II) are 6.1 V, 12 715 cd·m⁻² and 3.2 cd·A⁻¹, respectively. The enhancement of brightness and efficiency is 69% and 56%, respectively. On the other hand, device performances decrease

gradually with further increasing processing time. The performance of the modified device and unmodified one is listed in Table 1. The V - J - B characteristics of the device ITO/SAM (38) h/NPB(50 nm)/Alq₃(60 nm)/LiF/Al and Device II (inset) are shown in Fig.4.

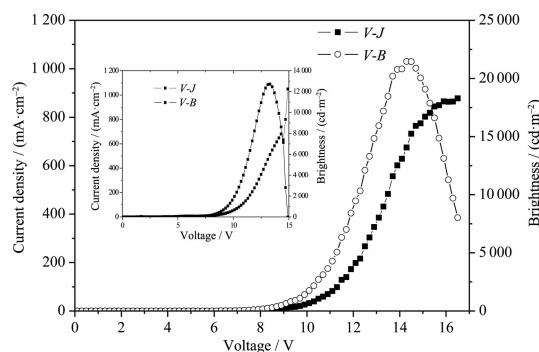


Fig.4 V - J - B characteristics of the device ITO/SAM (38) h/NPB (50 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al; Inset (the V - J - B characteristics of the Device II)

Fig.5 shows the current density-efficiency curves of the devices. The devices with ITO/SAM anodes show much higher efficiency than the unmodified ones. The modified ones could endure higher current density, exhibiting relatively high efficiency under high current density, e.g., the efficiency of the device

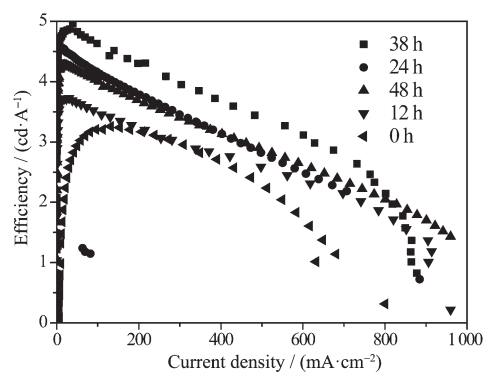


Fig.5 Efficiency-Current density characteristics of the modified (SAM processing 12~48 h) and unmodified (0 h) devices

Table 1 Performance of the devices with modified ITO (SAM processing time from 12 to 48 h) and unmodified ITO as anodes

		V_{on} / V	$B_{max} / (cd \cdot m^{-2})$	$Eff_{max} / (cd \cdot A^{-1})$
Unmodified / h	0	6.1	12 715	3.2
SAM processing time / h	12	5.9	16 658	3.7
	24	5.5	15 429	4.6
	38	5.5	21 464	5.0
	48	5.3	18 383	4.3

based on ITO/SAM (38) h is $2.2 \text{ cd} \cdot \text{A}^{-1}$ at $800 \text{ mA} \cdot \text{cm}^{-2}$, while that of the unmodified device decreases to $0.3 \text{ cd} \cdot \text{A}^{-1}$. The brightness of the devices is enhanced dramatically also after the ITO modification with SAM, as shown in Fig.6.

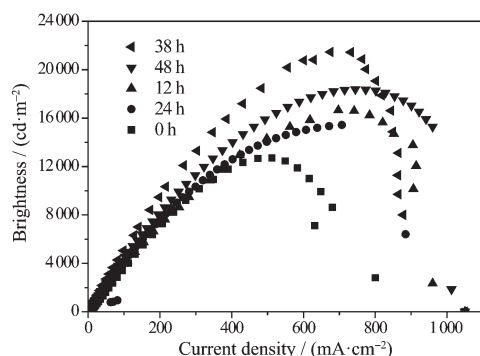


Fig.6 Brightness-Current density characteristics of the modified (SAM processing 12~48 h, 12~48 h) and the unmodified standard devices

Considering that Cz-Si contains the carbazole unit, the SAM is speculated to possess hole transporting ability. The other two devices were fabricated with thinner NPB layers:

Device III: ITO/SAM (40 h) /NPB (40 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al

Device IV: ITO/NPB (40 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al

The main performance of these two devices is listed in Table 2. As our speculation, the Cz-Si SAM layer could act as HTL partly, leading to relatively favorable performances when the thickness of NPB is appropriately reduced. On the other hand, the unmodified counterpart displays obviously lower EL performances. The brightness and efficiency of Device III improve 256% and 140% respectively, when compared with that of Device IV.

Fig.7 shows the efficiency-current density curves of Device III and Device IV. The efficiency of Device III increases rapidly under low current density, indicating high combination probability of holes and electrons, while the efficiency of Device IV increases

slowly under the same condition. This result indicates that the SAM modification improves the carrier balance of the device effectively.

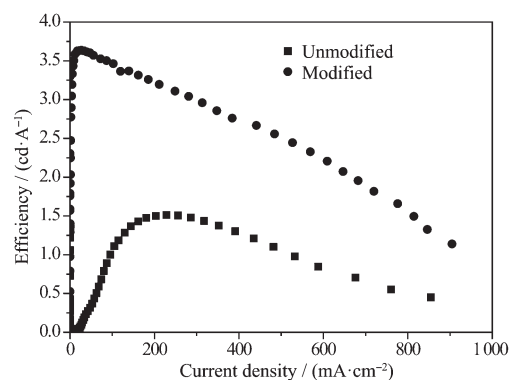


Fig.7 Efficiency-Current density characteristics of the Device III (modified) and Device IV (unmodified) devices

The operation stability of the device could also be improved remarkably after ITO SAM modification. The device with ITO/SAM anode shows better stability than the unmodified counterpart when the devices are without encapsulation in the ambient atmosphere for 30 d. Fig.8 shows the V - J - B curves of Device V and that of its counterpart Device VI (inset). The structures of these two devices are listed as follows:

Device V: ITO/SAM (60 h)/NPB (40 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al

Device VI: ITO/NPB (40 nm)/Alq₃ (60 nm)/LiF (1.0 nm)/Al

As shown in Fig.8, the V - J - B characteristics of Device V keep well. The maximum current density (J_{\max}) and B_{\max} of Device V maintain at about $600 \text{ mA} \cdot \text{cm}^{-2}$ and $5\,300 \text{ cd} \cdot \text{m}^{-2}$ respectively. In contrast, the J_{\max} and B_{\max} of Device VI decline to $50 \text{ mA} \cdot \text{cm}^{-2}$ and $520 \text{ cd} \cdot \text{m}^{-2}$ respectively, and the device should be operated at high working voltage. This result indicates that the SAM modification could improve the device stability effectively, which might be ascribed to the elimination of disparity in surface energies at the ITO/HTL interface after SAM modification.

Table 2 Performance of device III (modified) and Device IV (unmodified)

	V_{on} / V	$B_{\max} / (\text{cd} \cdot \text{m}^{-2})$	$\text{Eff}_{\max} / (\text{cd} \cdot \text{A}^{-1})$
Device III (modified)	4.7	20 995	3.6
Device IV (unmodified)	4.9	5 893	1.5

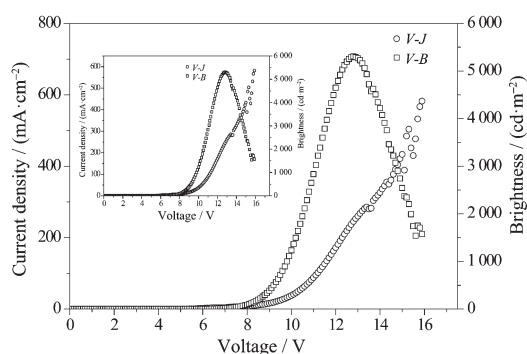


Fig.8 V - J - B characteristics of Device V and Device VI (inset) after being kept in air for 30 d

The effect of the SAM on the device performance could be understood in terms of hole injection barrier, interface integrity and surface roughness. Firstly, from the view of the energy level, insertion of Cz-Si SAM layer between ITO and HTL will increase the hole injection barrier, because the HOMO level of Cz-Si is lower than that of NPB, which results in lower hole injection efficiency. On the other hand, the image force^[26] resulted from molecular SAM could enhance the hole injection efficiency. These two reverse effects exist simultaneously within the heterostructure, and the observed result is the competing consequence. From our results, the image force effect is stronger than that of the energy level, because the V_{on} of the devices decreases a little. But the superiority is small because the decrease of V_{on} of the device is small after modification (Table 1 and 2). Secondly, the chemically grafted SAM film could eliminate the disparity at the ITO/HTL interface, to improve the growth quality of the organic film and the stability of the device. Our aqueous contact angle results show that the ITO/SAM is nearly hydrophobic as the organic HTL (e.g., NPB^[22]). This enhanced interfacial integrity will benefit for hole injection and film uniformity, offering better device performance and stability. Thirdly, the surface of ITO could be smoothed after modification with SAM, leading to pinhole-free organic functional layers and reduced leak current. Furthermore, the SAM layer could prevent In^{3+} ion migration from ITO anode to the emitting layer^[27], improving the device performance. Although enhanced device performance is achieved by

ITO SAM modification, the quality of the SAM should be controlled rationally. Because the HOMO level of Cz-Si is relatively low, Cz-Si layer with excessive thickness (e.g., formation of multilayer due to self-condensation from too long SAM processing time) would result in decline of enhancement rate for the efficiency and the brightness of the device as reflected in Table 1. The performance of the devices declines after the SAM processing time exceeding 38 h, which might be attributed to the over thickness of the Cz-Si film on the ITO surface.

3 Conclusions

A novel organosilane Cz-Si was designed, synthesized and used for the active material for SAM modification on ITO surface. Due to the stability of Cz-Si, the processing of the SAM modification is simple and convenient, which could be carried out in ambient atmosphere without any protection. By controlling the formation of the SAM layer, a series of electroluminescent devices with ITO/SAM or bare ITO as anodes were fabricated, and the devices with ITO/SAM anodes show remarkable improved performance when compared with their unmodified counterparts. Best device performance is achieved when ITO/SAM (38 h) is used as anode, and the V_{on} , B_{max} and Eff_{max} of this device are 5.5 V, 21 464 $cd \cdot m^{-2}$ and 5.0 $cd \cdot A^{-1}$, respectively. The device stability is improved also after ITO modification with SAM. The device with ITO/SAM anode exhibits better performance than its unmodified counterpart.

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References:

- [1] Kim J S, Cacialli F, Cola A, et al. *Appl. Phys. Lett.*, **1999**, **75**:19-21

- [2] Li C N, Kwong C Y, Djurišić A B, et al. *Thin Solid Films*, **2005**,**477**:57-62
- [3] Helander M G, Wang Z B, Qiu J, et al. *Science*, **2011**,**332**: 944-947
- [4] Lee S T, Wang Y M, Hou X Y, et al. *Appl. Phys. Lett.*, **1999**, **74**:670-672
- [5] Cui J, Huang Q, Veinot J G C, et al. *Adv. Mater.*, **2002**,**14**: 565-569
- [6] Brown T M, Kim J S, Friend R H, et al. *Appl. Phys. Lett.*, **1999**,**75**:1679-1681
- [7] Koch N, Kahn A, Ghijsen J, et al. *Appl. Phys. Lett.*, **2003**, **82**:70-72
- [8] Tang J X, Li Y Q, Hung L S, et al. *Appl. Phys. Lett.*, **2004**, **84**:73-75
- [9] Hsiao C C, Chang C H, Jen T H, et al. *Appl. Phys. Lett.*, **2006**,**88**:033512
- [10] Luo J X, Xiao L X, Chen Z J, et al. *Appl. Phys. Lett.*, **2008**, **93**:133301
- [11] Zhang H M, Fu Q, Zeng W J, et al. *J. Mater. Chem. C*, **2014**,**2**:9620-9624
- [12] Zhou L, Zhuang J Y, Tongay S, et al. *J. Appl. Phys.*, **2013**, **114**:074506
- [13] Chu T Y, Chen J F, Chen S Y, et al. *Appl. Phys. Lett.*, **2006**,**89**:053503
- [14] Hotchkiss P J, Jones S C, Paniagua S A, et al. *Acc. Chem. Res.*, **2012**,**45**:337-346
- [15] Kim D H, Chung C M, Park J W, et al. *Ultramicroscopy*, **2008**,**108**:1233-1236
- [16] Jonathan G C, Veinot, Marks T J. *Acc. Chem. Res.*, **2005**, **38**:632-643
- [17] Huang Q, Li J, Evmenenko G A, et al. *Chem. Mater.*, **2006**, **18**:2431-2442
- [18] Huang Q, Evmenenko G A, Dutta P, et al. *J. Am. Chem. Soc.*, **2005**,**127**:10227-10242
- [19] Yan H, Lee P, Armstrong N R, et al. *J. Am. Chem. Soc.*, **2005**,**127**:3172-3183
- [20] Huang Q, Cui J, Yan H, et al. *Appl. Phys. Lett.*, **2002**,**81**: 3528-3530
- [21] Huang Q, Evmenenko G, Dutta P, et al. *J. Am. Chem. Soc.*, **2003**,**125**:14704-14705
- [22] Wu Q H. *Crit. Rev. Solid State*, **2013**,**38**:318-352
- [23] HUANG Chun-Hui(黄春辉), LI Fu-You(李富友), HUANG Wei(黄维). *Introduction to Organic Light-emitting Materials and Devices*(有机电致发光材料与器件导论). Shanghai: Fudan University Press, **2005**:111
- [24] Cheng G, Zhang Y F, Zhao Y, et al. *Appl. Phys. Lett.*, **2005**,**87**:013506
- [25] Shi J and Tang C W. *Appl. Phys. Lett.*, **2002**,**80**:3201-3203
- [26] Tütis E, Bussac M N, Zuppiroli L. *Appl. Phys. Lett.*, **1999**, **75**:3880-3882
- [27] Wong K W, Yip H L, Luo Y, et al. *Appl. Phys. Lett.*, **2002**, **80**:2788-2790